

HU-ACE NEWS LETTER

Advanced Core for Energetics, Hiroshima University

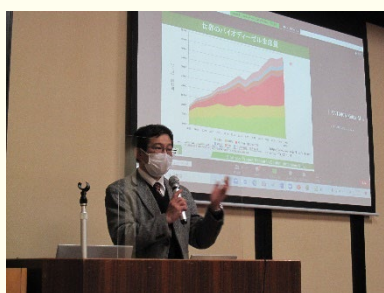
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Activities of the Core

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| Mar. 6, 2023 | The 27th Hiroshima University Biomass Project center Symposium (co-organized). |
| Mar. 8, 2023 | The 77th HU-ACE Steering Committee Meeting. |
| Mar. 13, 2023 | The 9th Hiroshima University Biomass Premium Evening Seminar (co-organized). |
| Mar. 20, 2023 | Post-Symposium the 3rd Anniversary of the Graduate School of Advanced Science and Engineering (co-organized). |

Biomass Symposium was held.

On March 6, 2023, the co-organized Biomass Symposium “Workshop: Biofuels from the basics” was held as a hybrid event. This symposium was held face-to-face for the first time in three years. After greetings from the head of HU-ACE, Prof. Yukihiro Matsumura, four lectures were given by Prof. Yutaka Nakashimada, Prof. Yukihiro Matsumura, Prof. Misturu Aoyagi, and Prof. Yoshimitu Uemura. The lectures were given with the aim of explaining the basics of biofuels in an easy-to-understand manner, introducing research seeds, and introducing the development and dissemination of biomass utilization technologies. After the lecture, there was a Q&A session and a lively exchange of opinions between online participants and on-site participants. It was a meaningful symposium that explained the current situation, cost issues, and progress in technological development in an easy-to-understand manner. We would like to thank everyone who was involved.



Related events

The 7th International Symposium on Fuels and Energy (ISFE 2023) will take place on July 3-4, 2023. We are preparing for a hybrid format event, including on-site participation at the Higashihiroshima Art and Culture Hall "Kurara", and online participation via ZOOM. Please visit the following ISFE 2023 website for more details.

<https://symposium2023.isfe.hiroshima-u.ac.jp>



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Research Topics

Utilization of functional property of alkali metal — Innovative NH₃ synthesis H₂ production —

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Research fields: Material science, Material / Energy conversion

Keywords: Nitride synthesis, H₂ storage, H₂ production



Abstract

Background

To suppress CO₂ emissions using fossil fuels and overcome energy issues in the future, enhancement of natural energy utilization is necessary. Various conversion techniques, which produce H₂ as energy media and energy carriers, such as NH₃ with high energy density, should be developed to establish energy systems based on fluctuated and localized natural energy.

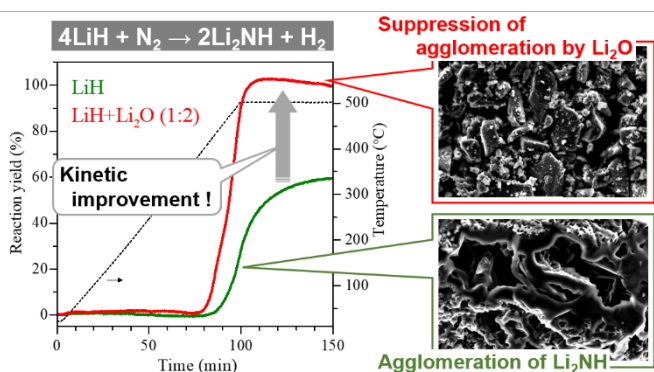
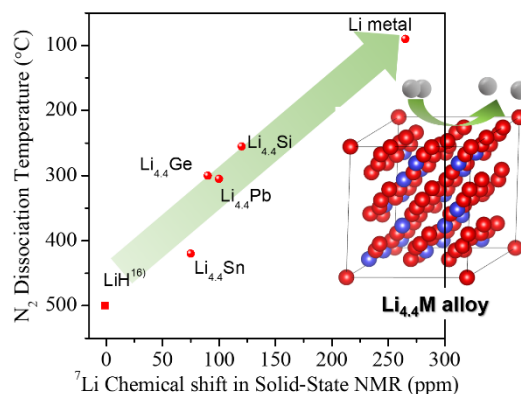
Methods

We focus on functional properties of alkali metals such as high reactivity and low melting points, and propose innovative material conversion techniques by controlling the above functions. In particular, we are experimentally studying small-scale / distributed types of ammonia (NH₃) synthesis systems and thermochemical hydrogen production techniques via water splitting which is operated below 500 °C and obtained from unused heat energy.

Results

We proposed a NH₃ synthesis technique based on functional properties such as high dissociation features and diffusion rate of Li, and experimentally investigated the reaction properties. As a result, we have clarified that NH₃ can be produced at an ambient pressure by chemical looping composed of three reactions using Li alloys. Also, that the nitrogen dissociation properties are related to the chemical state of Li in the alloys (upper figure).

The chemical looping using LiH, can also generate NH₃ at an ambient pressure through 2-step exothermic reactions. The main issue of this process is poor kinetics, and we have found through detailed analyses that the melting and condensation of intermediate phase during the reaction slows reactions. The mixing of lithium oxide with LiH as a scaffold to suppress the condensation drastically improves the kinetics (lower figure)



References

- [1] Shinzato, K.; Tagawa, K.; Tsunematsu, K.; Gi, H.; Singh P.K.; Ichikawa, T.; Miyaoka, H.: Systematic Study on Nitrogen Dissociation and Ammonia Synthesis by Lithium and Group 14 Element Alloys. *ACS Appl. Energy Mater.* **5**, 4765 (2022).
- [2] Tagawa, K.; Gi, H.; Shinzato, K.; Miyaoka, H.; Ichikawa, T.: Improvement of Kinetics of Ammonia Synthesis at Ambient Pressure by the Chemical Looping Process of Lithium Hydride. *J. Phys. Chem. C.* **126**, 2403 (2022).